3,4-Epoxy-8,9-dihydropiplartine. A New Imide from Piper verrucosum

Navindra P. Seeram,[†] Patrick A. Lewis,[‡] and Helen Jacobs^{*,†}

Department of Chemistry and Department of Botany, University of the West Indies, Mona, Kingston 7, Jamaica

Stewart McLean, William F. Reynolds,* Li-Lin Tay, and Margaret Yu

Department of Chemistry, University of Toronto, Toronto, Canada M5S 1A1

Received October 31, 1995[®]

The structure of a new compound isolated from the leaves and twigs of *Piper verrucosum* Sw. has been established as 3,4-epoxy-8,9-dihydropiplartine (**1**) by spectroscopic methods.

As part of our ongoing work on the phytochemistry of selected species of the endemic Jamaican flora, we have fractionated extracts of the leaves and twigs of *Piper verrucosum* Sw. (Piperaceae), a tree of relatively infrequent occurrence in central Jamaica.¹

The ground leaves and twigs were extracted with hexanes. The concentrated extract, upon repeated column chromatography and crystallization yielded only 3,4-epoxy-8,9-dihydropiplartine (1) (0.0003%). Fractionation and crystallization of the Me₂CO extract obtained from the marc afforded 3,4,5-trimethoxyben-zenepropanoic acid (0.0015%), identified from spectral data and comparison of the mp with the literature value.²



Compound 1, C₁₇H₂₁NO₆, mp 85-87 °C, [α]_D-121.6°, exhibited bands in its IR spectrum that indicated an aromatic ring (1596, 1548, 1473 cm⁻¹) and two carbonyl groups (1726, 1698 cm⁻¹). The ¹H-NMR spectrum displayed peaks characteristic of a trimethoxybenzenepropanoyl moiety.³ These consisted of coupled triplets for the C-8 and C-9 methylenes (δ 3.23 and δ 2.91, each 2H, J = 7.4 Hz); signals for three methoxyl groups, two of which are equivalent (δ 3.85, 6H, and δ 3.82, 3H), and a shielded aromatic singlet integrating for two protons (δ 6.46). The integration of the latter three signals demonstrated that the aromatic ring is symmetrically substituted as shown. The corresponding carbon signals (see Table 1) were assigned with the aid of HMQC spectra. Cross peaks in the HMBC spectra between both pairs of methylene protons (δ 3.23 and δ 2.91) and a carbon at δ 174.6 established this as the carbonyl of the substituted benzenepropanoyl group. Fragment ions in the EIMS at m/z 222, 194, 181, and 179 were consistent with the presence of this group.⁴

 1 H and 13 C signals for the C₅H₆NO₂ residue indicated an amidic carbonyl, a disubstituted epoxide, and two methylene groups. HMBC cross peaks between the

Table 1.	NMR Data o	of Compound	1 [δ	(ppm).	CDCl ₃]
	I THIN D GLCG U	. compound	- 10	(PP)	020131

position	$\delta_{ m C}$	$\delta_{ m H}{}^{a}$	$HMBC^{b}$
2	169.6		3.56
3	52.3	3.56 (1H, d, J =	3.70, 2.42
		4.1 Hz)	
4	53.4	3.70 (1H, dd, $J =$	4.34, 2.42
		4.1, 5.8 Hz)	
5	23.8	2.42 (1H, m)	4.34, 3.70
		2.00 (1H, ddd, $J =$	
		15.0, 5.8, 13.2 Hz)	
6	35.6	4.34 (1H, dddd, $J =$	2.00
		13.5, 5.8, 1.5, 1.5 Hz)	
		3.20 (1H, m)	
7	174.6		3.23, 2.91
8	41.2	3.23 (2H, t, J = 7.4 Hz)	2.91
9	31.2	2.91 (2H, t, $J = 7.4$ Hz)	6.46, 3.23
10	136.6		6.46, 3.23,
			2.91
11, 15	105.4	6.46 (2H, s)	6.46, 2.91
	(double)		
12, 14	153.1		6.46, 3.85
	(double)		
13	136.3		3.82
16, 18	56.0	3.85 (6H, s)	
	(double)		
17	60.8	3.82 (3H, s)	

 a Values were established by HMQC. b Protons that correlate with carbon resonance.

carbonyl carbon (δ 169.6) and the epoxymethine proton at δ 3.56 demonstrated that the epoxy and carbonyl functionalities are vicinal. Analysis of HMQC, COSY, and HMBC data enabled the complete assignment of the signals for this residue (see Table 1) leading to its formulation as 3,4-epoxy-2-oxopiperdinyl.

The relative stereochemistry was determined from the magnitude of the vicinal coupling between H-3 and H-4 and the results of a NOESY experiment. The *J* value $(J_{3-4} = 4.1 \text{ Hz})$ was exactly as expected for a cis coupling with a near-zero dihedral angle in a 1,2-epoxycyclohexane.⁵ A trans coupling in the same system should have a coupling of ~0 Hz. The NOESY spectrum showed a strong cross peak between H-3 and H-4, again consistent with a cis stereochemistry. H-4 showed typical gauche couplings along with NOESY cross peaks to both C-5 protons, suggesting that it is in a pseudoequatorial orientation in the six-membered ring. This ring will obviously be distorted from a chair form by the carbonyl and epoxide functionalities.

Compound **1** is a new amide alkaloid and may be regarded as the 3,4-epoxy-8,9-dihydro derivative of the known compound piplartine, which occurs in the roots of the Indian plant *Piper longum* L.⁶ A similar compound in which the heterocyclic residue is 3,4-didehy-

[†] Department of Chemistry.

 [‡] Department of Botany.
 [®] Abstract published in Advance ACS Abstracts, March 15, 1996.

dro-2-oxopyrrolidinyl has been isolated from *Piper demeraranum* (Miq.) C. DC. of Trinidad.³

Epoxides are not common in the *Piper* genus and have been reported only in the Old World species *P. polysyphonum* C. DC. of China and *P. hookeri* Hook, *P. cubeba* (Miq.), and *P. brachystachum* Wall of India. These plants produce the chorismate-derived cyclohexane epoxides, crotepoxide, and related compounds.^{7–9}

Experimental Section

General Experimental Procedures. Melting points were determined on a Thomas-Hoover capillary melting point apparatus. The optical rotation was measured on a Perkin-Elmer 24MC polarimeter. IR spectra were taken on a Perkin-Elmer 735B spectrophotometer as KBr pellets, and the UV spectrum was run on a Pye Unicam 8800 spectrophotometer. MS were obtained on a VG 70-250S mass spectrometer. NMR spectra were determined using a Bruker ACE 200 and a Varian UNITY-500 spectrometer with TMS as internal standard. Column chromatography utilized SiO₂, Mallinkrodt, 230–400 mesh.

Plant Material. The leaves and twigs of *P. verrucosum* Sw. (Piperaceae) were collected near Quickstep, Trelawny, Jamaica, in March 1995. A voucher specimen (no. 33,621) is lodged in the Herbarium at the University of the West Indies, Mona, Jamaica.

Extraction and Isolation. Dried leaves and twigs (1.64 kg) were ground and extracted by cold percolation with hexanes. Evaporation yielded a gum (37 g), a portion (12 g) of which was chromatographed (gradient elution with Me₂CO-hexanes). The residue (1.5 g) from the 20% Me₂CO fractions was chromatographed in 5% Me₂CO-hexanes to yield pure **1** (157 mg).

Extraction with cold Me₂CO of the marc from the hexane extract and evaporation of the solvent gave a

gum (57 g), a portion (12 g) of which was chromatographed (Me₂CO-hexanes gradient). The fractions eluted with 50% Me₂CO were rechromatographed, again using a Me₂CO-hexanes gradient. 3,4,5-Trimethoxybenzenepropanoic acid (500 mg) was obtained from fractions eluted with 25% Me₂CO-hexanes.

3,4-Epoxy-8,9-dihydropiplartine (1): Recrystallized from Me₂CO-ligroin, mp 85–87 °C; $[\alpha]_D$ –121.6° (*c* 0.020, CHCl₃); IR *v* max 1726, 1698, 1596, 1548, 1473 cm⁻¹; UV λ max (EtOH) (log ϵ) 260 (2.92) nm; EIMS *m*/*z* [M]⁺⁺ 335 (95), 222 (100), 194 (66), 181 (69), 179 (66); HRMS *m*/*z* found 335.1384, calcd 335.1369 for C₁₇H₂₁O₆; ¹H and ¹³C-NMR data, see Table 1.

Acknowledgements. Financial support from the University of the West Indies Research and Publications Fund Committee and the Natural Sciences and Engineering Research Council of Canada is gratefully acknowledged. We thank the staff of the Conservation Data Centre UWI-Mona for assistance with plant collection.

References and Notes

- (1) Adams, C. D. *Flowering Plants of Jamaica*; The University Press: Glasgow, 1972; p 213.
- (2) Beilsteins Handbuch der Organischen Chemie, Springer-Verlag: Berlin, 1971; Vol. 10, p 2120.
- (3) Maxwell, A.; Rampersad, D. J. Nat. Prod. 1989, 52, 891-892.
- (4) Chatterjee, A.; Dutta, M. Tetrahedron 1967, 23, 1769-1781.
- (5) Tori, K.; Komeno, T.; Nakagawa, T. J. Org. Chem. 1964, 29, 1136-1141.
- (6) Joshi, B. S.; Kamat, V. N.; Saksena, A. K. Tetrahedron Lett. 1968, 2395–2400.
- (7) Ma, Y.; Hau, G.-Q. J. Chinese Pharm. Sci. 1993, 2, 97-101.
- (8) Singh, J.; Dhar, K. L.; Atal, C. K. Tetrahedron 1970, 26, 4403– 4406.
- (9) Taneja, S. C.; Koul, S. K.; Pushpangadan, P.; Dhar, K. L.; Danewski, W. M.; Schilf, W. Phytochemistry 1991, 30, 871–874.

NP9602168